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Note

The synthesis of 2,3-didehydro-2,4-dideoxy-4-guanidinyl-N-acetylneuraminic acid: a potent influenza virus sialidase inhibitor

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The biochemistry and synthesis of sialidase inhibitors, in particular influenza virus sialidase inhibitors, has been of great interest to us [1–3]. The synthesis of a number of 5-acetamido-2,6-anhydro-3,5-dideoxy-D-glycero-D-galacto-non-2-enonic acid (Neu5Ac2en, 1) analogues has received considerable attention over the past decade [4–6]. These compounds are thought to be transition-state analogues of the enzyme reaction [1,2]. We have recently reported the design and biological evaluation of 5-acetamido-4-amino-2,6-anhydro-3,4,5-trideoxy-D-glycero-D-galacto-non-2-enonic acid (4-amino-Neu5Ac2en, 2) and 5-acetamido-2,6-anhydro-4-guanidino-3,4,5-trideoxy-D-glycero-D-galacto-non-2-enonic acid (4-guanidino-Neu5Ac2en, 3) as influenza virus sialidase inhibitors [7]. Both of these compounds, in particular 3, have been shown, not only to be potent inhibitors of influenza virus sialidase and influenza virus replication, but also of the virus in a number of animal models [7].

The synthetic strategy that we adopted for the preparation of 3 required the introduction of nitrogen at C-4 of 1. We have previously reported the preparation of 4 in good yield, and this was achieved by treatment of 1 with BF₃ Et₂O, followed by reaction with azide [8]. The synthesis of 3 was successfully completed as detailed in Scheme 1. Thus, reduction of 4 was best achieved by hydrogenation at atmospheric pressure in the presence of Pd-C (10%) in toluene, methanol, and glacial acetic acid for 1 h, which afforded, following work-up and purification, methyl 5-acetamido-7,8,9-tri-O-acetyl-4-amino-2,6-anhydro-3,4,5-trideoxy-D-glycero-D-galacto-non-2-enonate (4-amino-Neu5,7,8,9Ac₄2en1Me, 5) in 72% over-

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all yield. These reaction conditions for the reduction of the azide to 5 were found to be critical because of the possibility of unwanted side reactions, for example, acetate migration or even over-reduction to give the corresponding 4-acetamido and 2,3-dideoxy saturated analogues, respectively. 4-Amino-Neu5,7,8,9Ac₄2en1Me (5) was readily converted to 6 by sequential treatment with Amberlite-IRA 400 (OH⁻) resin and aqueous sodium hydroxide, followed by neutralisation with Dowex-50W \times 8 (H⁺) resin and lyophilisation in an overall isolated yield of 66% from 4. We have found that our strategy for the preparation of 6 was a significant improvement on a previously reported triphenylphosphine reduction method which gave the corresponding free acid 2 in a 31% overall isolated yield from 4 (ref 5). 6 was cleanly guanidinated by treatment with aminoiminomethanesulfonic acid [9] in the presence of aqueous potassium carbonate with stirring at 30–40°C for 8 h and subsequently at room temperature until the reaction was determined by ¹H NMR spectroscopy (D₂O) to be complete. After workup and purification, 3 was isolated in 57% yield.

Compound 3, existing presumably as a zwitterion at physiological pH, has been determined to be a slow-binding inhibitor of influenza virus sialidase with a binding affinity of 10^{-10} M (refs. 3 and 7). Further studies, including clinical trials, are currently underway to determine the potential of 3 as an anti-influenza drug candidate.

1 $R \approx OH$, R' = NHAc

 $2 R = NH_2$, R' = NHAc

 $3 R = NH(C=NH)NH_2, R' = NHAc$ 10

HOP HOP NH2-Pd/C

MeOH/Toluene/AcOH

$$R = Me, R' = NHAc$$
 $R = Me, R' = NHAc$
 $R'' = Ac$

1) saponification ('OH)

5

1) saponification (H*)

6

 $R' = NHAc$

1) aqueous K_2CO_{1}
 K_2CO_{2}
 K_3CO_{3}
 K_4
 K_5
 K_7
 K_7
 K_8
 K_9
 K

Scheme 1.

1. Experimental

Methyl 5-acetamido-7,8,9-tri-O-acetyl-4-amino-2,6-anhydro-3,4,5-trideoxy-D-glycero-D-galacto-non-2-enonate (5, 4-amino-Neu5,7,8,9Ac₄2en1Me).—To a solution of 4 (1 g, 2.19 mmol) in MeOH (38 mL) was added toluene (23 mL), Pd-C (10%) (190 mg), and acetic acid (0.2 g, 3.33 mmol). This mixture was hydrogenated at atmospheric pressure for 1 h and then filtered. The filtrate was evaporated to dryness, and the residue was subjected to flash chromatography (silica gel, 5:2:1 EtOAc-2-propanol-water) to afford pure 5 (0.68 g, 72%); $\nu_{\rm max}$ (CHCl₃) cm⁻¹ 3400 (amine), 1740 (COOMe); ¹H NMR data (300 MHz, CDCl₃ + CD₃OD): δ 1.96, 2.06, 2.07, 2.10 (s, 12 H, 3 AcO, CH₃CONH), 3.81 (s, 3 H, CO₂CH₃), 3.92 (br t, 1 H, $J_{5,4}$ 10, $J_{5,6}$ 10 Hz, H-5), 4.17 (dd, 1 H, $J_{9,8}$ 7.2, $J_{9,9}$ 12.3 Hz, H-9'), 4.22 (br dd, 2 H, $J_{4,5}$ 10, $J_{4,3}$ 2.1, $J_{6,5}$ 10, $J_{6,7}$ 2.1 Hz, H-4,6), 4.71 (dd, 1 H, $J_{9,8}$ 2.6, $J_{9,9}$ 12.3 Hz, H-9), 5.31 (m, 1 H, $J_{8,7}$ 4.9, $J_{8,9}$ 2.6, $J_{8,9}$ 7.2 Hz, H-8), 5.45 (dd, 1 H, $J_{7,6}$ 2.1, $J_{7,8}$ 4.9 Hz, H-7), 5.97 (d, 1 H, $J_{3,4}$ 2.1 Hz, H-3); ¹³C NMR data (300 MHz, CDCl₃ + CD₃OD): δ 22.2-22.3 (3 CH₃CO₂), 24.3 (CH₃CONH), 50.2 (C-5), 52.4

(C-4), 54.0 (CO₂CH₃), 64.1 (C-9), 69.8, 73.2 (C-7, C-8), 78.5 (C-6), 114.5 (C-3), 145.6 (C-2), 164.0 (C-1), 172.2, 172.4, 172.8, 174.2 (3 CH₃CO₂, CH₃CONH); FABMS: 431 (M + 1)⁺, 414 (M⁺ – NH₂). Anal. Calcd for $C_{18}H_{26}N_2O_{10}$: C, 50.2; H, 6.1; N 6.5. Found C, 50.0; H, 6.0; N, 6.4.

5-Acetamido-2,6-anhydro-4-guanidino-3,4,5-trideoxy-D-glycero-D-galacto-non-2enonic acid (3, 4-guanidino-Neu5Ac2en).—To a solution of 5 (1 g, 2.32 mmol) in anhyd MeOH (65 mL) was added dried Amberlite IRA-400 (OH⁻) resin (3 g). The mixture was stirred for 3 h at room temperature and then filtered. The resin was washed with MeOH $(2 \times 30 \text{ mL})$, and the filtrate and washings were concentrated to dryness to afford a white solid. This material was then taken up in 0.15 M ag NaOH (15 mL), stirred at room temperature for 2 h, and finally the pH of the mixture was adjusted to 7.0-7.5 with Dowex-50W \times 8 (H⁺) resin. After filtration, the filtrate was lyophilised to afford compound 6 (0.67 g, 91.6%), which was very similar by ¹H NMR spectroscopy to the corresponding free acid 2 previously reported [5]; ν_{max} (CHCl₃) cm⁻¹ 3416 (amine), 1628 (COO⁻); ¹H NMR data (300 MHz, D₂O): δ 2.07 (s, 3 H, CH₃CONH), 3.59–3.70 (m, 2 H, H-7,9), 3.89 (dd, 1 H, $J_{9.8}$ 2.6, $J_{9.9}$ 11.8 Hz, H-9), 3.95 (m, 1 H, H-8), 3.99 (br d, 1 H, $J_{4,5}$ 10.6 Hz, H-4), 4.21 (br t, 1 H, $J_{5.4}$ 10.6, $J_{5.6}$ 10.6 Hz, H-5), 4.29 (br d, 1 H, $J_{6.5}$ 10.6 Hz, H-6), 5.65 (d, 1 H, J_{34} 1.9 Hz, H-3); ¹³C NMR data (300 MHz, D₂O): δ 22.2 (C-11), 46.8 (C-5), 50.1 (C-4), 62.2 (C-9), 68.0 (C-8), 69.8 (C-7), 75.2 (C-6), 101.8 (C-3), 150.2 (C-2), 168.8, 174.9 (C-1,10); FABMS: 313 $(M+1)^+$.

To a well-stirred solution of compound 6 (6 g, 19.23 mmol) in water (100 mL) were added aminoiminomethanesulfonic acid [9] (8.3 g, 66.94 mmol) and potassium carbonate (9.38 g, 67.4 mmol) over a period of 8 h while maintaining a temperature of 30-40°C. The turbid mixture was then stirred at room temperature overnight until ¹H NMR (in D₂O) showed that the reaction was complete. The mixture was diluted with 100 mL of water and filtered. The filtrate was then treated with Amberlite IR-120 (H⁺) resin (400 mL), and the resin was washed with water (4 L), then eluted with 0.6 M ag Et₃N. The isolated crude triethylammonium 4guanidino-Neu5Ac2en was taken up in water and reconcentrated five times (200 mL/cycle), which afforded crude 3 (7.6 g). This material was then taken up in water (100 mL), heated to 40-50°C and filtered. The filtrate was diluted with 2-propanol and warmed with stirring upon which the mixture become turbid. The turbid suspension was then further stirred for 16 h at room temperature and 4-guanidino-Neu5Ac2en (3) was isolated as colourless crystals (4.04 g, 57%). $[\alpha]_D^{20} + 40.9^{\circ} (c \ 0.9, H_2O);$ ¹H NMR data (300 MHz, D₂O): δ 1.92 (s, 3 H, AcO), 3.53 (dd, 1 H, $J_{9,8}$ 6.2, $J_{9,9}$ 11.9 Hz, H-9'), 3.56 (dd, 1 H, $J_{7,8}$ 9.0, $J_{7,6}$ 1.1 Hz, H-7), 3.77 (dd, 1 H, $J_{9.8}$ 2.7, $J_{9.9'}$ 11.9 Hz, H-9); 3.84 (ddd, 1 H, $J_{8.9}$ 2.7, $J_{8.9'}$ 6.2, $J_{8.7}$ 9.0 Hz, H-8), 4.11 (dd, 1 H, $J_{5.4}$ 9.2, $J_{5.6}$ 10.5 Hz, H-5), 4.27 (dd, 1 H, $J_{6.7}$ 1.1, $J_{6.5}$ 10.5 Hz, H-6), 4.34 (dd, 1 H, $J_{4,3}$ 2.3, $J_{4,5}$ 9.2 Hz, H-4), 5.51 (d, 1 H, $J_{3,4}$ 2.3 Hz, H-3); 13 C NMR (D₂O): δ 23.1 (CH₃CONH), 48.9 (C-5), 52.2 (C-4), 64.1 (C-9), 69.2 (C-7), 70.9 (C-8), 76.5 (C-6), 104.9 (C-3), 150.3 (C-2), 158.1 (C-10), 170.1, 175.9 (C-1, CH₃CONH); mp 256°C (dec); FABMS: 333 $(M + 1)^+$; calculated 333.31. Anal. Calcd for C₁₂H₂₀N₄O₇ · 1.5H₂O: C 40.11; H 6.45; N 15.59. Found: C 40.1; H 6.7; N 15.7.

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